

AP Biology - Identifying Intermolecular Forces

Name = _____

Indicate the **intermolecular &/or intramolecular force(s)** that can exist between the following particles and where they are located.

	London Dispersion Forces <i>(form between multiple copies of this/these substance(s) in an area)</i> <i>*Identify between which substances these form.</i>	Dipole - Induced Dipole Forces <i>(form between multiple copies of this/these substance(s) in an area)</i> <i>*Identify between which substances these form.</i>	Dipole - Dipole Forces <i>(form between multiple copies of this/these substance(s) in an area)</i> <i>*Identify between which substances these form.</i>	Hydrogen Bonds: Special Dipole-Dipole Forces <i>(form between multiple copies of this/these substance(s) in an area)</i> <i>*Identify between which substances these form.</i>	Ion - Dipole Forces <i>(form between multiple copies of this/these substance(s) in an area)</i> <i>*Identify between which substances these form.</i>	Ionic Bonds Form Between the Ions Present in this/these Substance(s) <i>*Identify between which substances these form.</i>	Nonpolar Covalent Bonds <i>(Identify which substance this applies to if more than 2 substances exist)</i> <i>*Identify between which atoms these form.</i>	Polar Covalent Bonds <i>(Identify which substance this applies to if more than 2 substances exist)</i> <i>*Identify between which atoms these form.</i>
CaOH & H₂O			X (between H ₂ O & H ₂ O;)	X (between H ₂ O & H ₂ O;)	X (between Ca ²⁺ & H ₂ O; OH ⁻ & H ₂ O;)	X (between Ca ²⁺ & OH ⁻)		X (in the H ₂ O between the H and the O)
O₂								
HCl (when not dissociated)								
C₃H₈								
CO₂								
NH₃								
HF & H₂O (when not dissociated)								
NaCl (not in water)								
H₂O & CH₄								
KOH (when placed in water)								
NO								
C₂H₅OH								