Write ionic formulas for the following compounds:

a. Sodium Acetate
b. Tin (II) Chloride
c. Calcium Hydroxide
d. Zinc Sulfite
e. Ammonium Sulfate
f. Manganese (II) Hypochlorite
g. Copper (I) Nitrite
h. Silver Cyanide
i. Nickel (II) Chlorite
j. Lead (II) Nitrate
k. Sodium Chloride
l. Lithium Fluoride
m. Potassium Sulfide
o. Aluminium Oxide

m. $PbSO_4$
n. $Mg_3(PO_4)_2$
o. Na_2CO_3
p. $NaHCO_3$
q. KCN
r. $KMnO_4$