

1) Name the following ionic compounds:

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|----------------------------|---------------------------|
| a. LiCl | g. AgNO_3 |
| b. Mg(OH)_2 | h. NH_4Cl |
| c. K_3P | i. CuCl_2 |
| d. Fe_2O_3 | j. SnCl_2 |
| e. FeO | k. PbO_2 |
| f. ZnCl_2 | l. AlCl_3 |

2) Write ionic formulas for the following compounds:

- Sodium Acetate
- Tin (II) Chloride
- Calcium Hydroxide
- Zinc Sulfite
- Ammonium Sulfate
- Manganese (II) Hypochlorite
- Copper (I) Nitrite